

Progressive changes in surface structure and electronic properties on Si(001) surface by CaF₂ adsorption

A. Z. Alzahrani and D. Usanmaz
Physics Department, Faculty of Science, King Abdulaziz University, P.O. Box 80203, Jeddah 21589, Saudi Arabia
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Ab initio calculations, based on pseudopotentials and density functional theory, have been performed to investigate the atomic geometry and electronic structures of calcium fluoride (CaF₂) on the Si(001) surface. We have considered the experimentally observed (2×1) and (3×1) reconstructions with different bonding configurations of the CaF₂ molecule on the Si(001) surface. Our total energy calculations suggest that the (3×1) structure is slightly more preferable than the (2×1). The key structural parameters and electronic surface properties of their most stable structures have been discussed. In contrast to the experimental results, the most stable structures contain an appreciable bonding nature between the surface Si and F atoms, however, no real bonding between Si and Ca atoms is indicated.

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Article Outline

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